

Name: \_\_\_\_\_

Calculate the pH of a 0.065 M solution of HCl.

Ans: pH = \_\_\_\_\_

Name: \_\_\_\_\_

Calculate the pH of a 0.0097 M solution of  $\text{HNO}_3$ .

Ans: pH = \_\_\_\_\_

Name: \_\_\_\_\_

Calculate the pH of a 0.0028 M solution of  $\text{HClO}_3$ .

Ans: pH = \_\_\_\_\_

Name: \_\_\_\_\_

Calculate the pH of a 0.00019 M solution of  $\text{HIO}_3$ .

Ans: pH = \_\_\_\_\_

Name: \_\_\_\_\_

Calculate the pH of a 0.66 M solution of  $\text{HBrO}_3$ .

Ans: pH = \_\_\_\_\_

Name: \_\_\_\_\_

Calculate the pH of a 0.067 M solution of HBr.

Ans: pH = \_\_\_\_\_

Name: \_\_\_\_\_

Calculate the pH of a 0.00098 M solution of HI.

Ans: pH = \_\_\_\_\_

Name: \_\_\_\_\_

Calculate the pH of a 0.00023 M solution of  $\text{HClO}_4$ .

Ans: pH = \_\_\_\_\_



Name: \_\_\_\_\_

Calculate the pH of a 1.4 M solution of  $\text{HBrO}_4$ .

Ans: pH = \_\_\_\_\_

Name: \_\_\_\_\_

Calculate the pH of a 0.0024 M solution of  $\text{HIO}_4$ .

Ans: pH = \_\_\_\_\_

copy 1: pH= 1.19

copy 2: pH= 2.01

copy 3: pH= 2.56

copy 4: pH= 3.73

copy 5: pH= 0.18

copy 6: pH= 1.17

copy 7: pH= 3.01

copy 8: pH= 3.63

copy 9: pH= -0.15

copy 10: pH= 2.62