Critical Item #5

NAME _____

Calculate the concentration (molarity) of a solution which has 25.0 g of NaCl in 250 mL of solution.

ANS _____

Critical Item #5

NAME _____

Calculate the concentration (molarity) of a solution which has 15.0 g of NaF in 500 mL of solution.

ANS _____

Critical Item #5

NAME _____

Calculate the concentration (molarity) of a solution which has 25.0 g of CH_3COOH in 250 mL of solution.

ANS _____

Critical Item #5

NAME _____

Calculate the concentration (molarity) of a solution which has 25.0 g of KCl in 250 mL of solution.

ANS _____

Critical Item #5

NAME _____

Calculate the concentration (molarity) of a solution which has 35.0 g of $CaCl_2$ in 250 mL of solution.

ANS _____

Critical Item #5

NAME _____

Calculate the concentration (molarity) of a solution which has 5.0 g of $NaNO_3$ in 250 mL of solution.

ANS _____