CHE 121 Critical Item 5 - Molarity - June 12, 2008	Copy 1
Critical Item #5	
NAME	
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Calculate the concentration (molarity) of a solution which has 25.0 g of NaCl in 250 mL of solution.

CHE 121 Critical Item 5 - Molarity - June 12, 2008	Copy 2
Critical Item #5	
NAME	

Calculate the concentration (molarity) of a solution which has 15.0 g of NaF in 500 mL of solution.



CHE 121 Critical Item 5 - Molarity - June 12, 2008	Copy 3
Critical Item #5	
NAME	
Calculate the concentration (molarity) of a solution which has 25.0 g of CH ₃ COOH in 250 solution.) mL of

ANS _____

CHE 121 Critical Item 5 - Molarity - June 12, 2008	Copy 4
Critical Item #5	
NAME	

Calculate the concentration (molarity) of a solution which has 25.0 g of KCl in 250 mL of solution.



CHE 121 Critical Item 5 - Molarity - June 12, 2008	Copy 5
Critical Item #5	
NAME	

Calculate the concentration (molarity) of a solution which has $35.0~{\rm g}$ of ${\rm CaCl_2}$ in $250~{\rm mL}$ of solution.



CHE 121 Critical Item 5 - Molarity - June 12, 2008	Copy 6
Critical Item #5	
NAME	
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Calculate the concentration (molarity) of a solution which has $5.0~{\rm g}$ of NaNO $_3$ in 250 mL of solution.

