Critical Question #3

Name _____
                             please print

Calculate the number of molecules in 225 mole of NH_3.

$(N_A = 6.022 \times 10^{23} \text{ mol}^{-1})$

_____ molecules

Critical Question #3

Name _____
                                            please print

Calculate the number of molecules in 125 mole of NH_3.

($N_A = 6.022 \times 10^{23}$ mol^{-1})

_____ molecules

Critical Question #3

Name _____
                                    please print

Calculate the number of molecules in 11.3 mole of NH_3.

(N_A = 6.022 x 10^{23} mol^{-1})

_____ molecules

Critical Question #3

Name _____

please print

Calculate the number of molecules in 650 mole of NH_3.

$(N_A = 6.022 \times 10^{23} \text{ mol}^{-1})$

_____ molecules

Critical Question #3

Name _____
                              please print

Calculate the number of molecules in 5.15 mole of NH_3.

$(N_A = 6.022 \times 10^{23} \text{ mol}^{-1})$

_____ molecules